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(56) References cited: :

- (54) Esters of (R)(-)-carnitine and acyl (R)(-)-carnitines with beta-hydroxybutyric acid and pharmaceutical compositions containing them for inhibiting neuronal degeneration, liver proteolysis and for the treatment of coma
- (57) The esters of (R)(-)-carnitine and acyl (R)(-)-carnitines with beta-hydroxybutiric acid in the form of pharmacologically acceptable salts of formula (I)

wherein X is the anion of a pharmacologically acceptable salt, e.g. chloride, bromide, orotate, acid aspartate, acid citrate, acid phosphate, acid fumarate, lactate, acid maleate, acid oxalate, acid sulfate and glucosepbosphate or in the form of inner salts of formula (1).

wherein R is hydrogen or a straight or branched acyl group having from 2 to 5 carbon atoms, such as e.g. acetyl, propionyl, nbusyryl, siebultyryl and isovaleryl, are active in inhibiting neuronal degeneration (as it occurs in Abzheimer's semile dementia and Parkinson's disease), liver proteolysis and in the treatment of corns.

#### Description

[0001] The present invention relates to the esters of R(-)-carnitine and acyl (R) (-)-carnitines with beta-hydroxybutyric acid in the form of their pharmacologically acceptable salts of formula (I)

wherein X is the anion of a pharmacologically acceptable acid e.g. chloride, bromide, orotate, acid aspartate, acid citrate, acid phosphate, acid fumarate, lactate, acid maleate, acid oxalate, acid sulfate and glucosephosphate, or in the form of inner salts of formula (I')

wherein R is a hydrogen or a straight or branched acyl group having from 2 to 5 carbon atoms, such as for instance acetyl, propionyl, n-butyryl, isobutyryl and isovaleryl.

[0002] These compounds are active in inhibiting neuronal degeneration (as it occurs in Alzheimer's senile dementia and Parkinson's disease) and liver proteolysis and in the treatment of coma.

[0003] The present invention also relates to orally or parenterally admnistrable pharmaceutical compositions for treating the foregoing pathologies, which comprise one of the compounds of formula (I) or (I') as active principle.

[0004] Esters of camitine with hydroxy-substituted saturated organic acids (e.g. 2-hydroxybutyric, 2-hydroxy-2-methylbutyric and 2-methyl-3-hydroxy propionic acid) are known already; see e.g. US patent 4,766,222 assigned to Sigma-Tau Industrie Farmaceutiche Riunite S.p.A. These compounds, however, are O-esters (i.e. esters on the camitine hydroxyl group) and endowed with pharmacological properties entirely different from and in no way related to the properties of the esters of the present invention.

[0005] Esters on the camitine carboxyl group are described in Z. Physiol. Chem., 295, 377, 1953 and Z. Physiol. Chem., 346, 314, 1966. These are, however, esters of camitine with allphatic alcohols such as methanol, ethanol and butanol or with aromatic alcohols such as bearval alcohol. not with hydroxy-acids.

[0006] The non-limiting examples that follow show the preparation of the esters of acyl (R)(-)-carnitine chloride with beta-hydroxybutyric acid via the synthesis scheme which is illustrated hereinbelow.

Example 1

[0007] Preparation of the ester of isovaleryl (R)(-)-carnitine chloride with (R,S)(±)-beta-hydroxybutyric acid (ST 687).

Step a:

[0008] Preparation of the benzyl ester of (R,S)(±)-beta-hydroxybutyric acid <u>1</u> (R,S)(±)-beta-hydroxybutyric acid sodium salt (1.2 g; 0.01 moles) was suspended in benzyl bromide (6 ml; 0.05 moles).

[0009] 18 crown - 6 (0.264 g) dissolved in 7 ml acetonitrile was added to the mixture.

[0010] The solution was partially concentrated under a nitrogen stream and then kept under stirring at 80°C for 90 minutes. Following cooling, hexane - HoO was added. The separated and dried organic phase was concentrated and then distilled under vacuum to remove the excess of heavyl bromide.

 $[0011] \ \ A \ solid \ residue \ (1.1 \ g) \ was \ obtained \ which \ was \ identified \ as \ the \ title \ product, \ yield \ 56\%, \ TLC \ CHCls9 \ - \ MetOH \ 1$ 

Rf = 0.8

Gas chromatography column HP1 25 m; 0.32 mm ID; 0.33 um film

Gas chrom thickness

carrier (He) flow rate: 1 ml/min Make up gas 40 ml/min

Splitting ratio 40 ml/min Injector 220°C

Detector(Fid) 280°C

T column 120°C for 3 minutes, 15°C/min 250°C

Rt = 9.36 product 1

Rt = 4.84 benzylbromide absent

NMR CDClsô7.3(5H,s,benzyl); 5.2(2H,s,CH<sub>2</sub>-benzyl); 4.2(1 H,m,CH);

2.8(1 H.s,broad OH); 2.5(2H,d,-CH2COO); 1.2(3H,d,CH3)

Step b:

[0012] Preparation of the acid chloride of isovaleryl  $R(\cdot)$ -camitine chloride 2 Thionyl chloride (7.7 ml; 0.1 moles) was added to isovaleryl  $(R)(\cdot)$ -camitine chloride (10 g; 0.035 moles).

[0013] The resulting mixture was kept at room temperature for 4 hours, then concentrated under vacuum to remove the excess of thionyl chloride. The residue was washed 3 times with anhydrous ethyl ether. The reaction raw product thus obtained was used in the following step without further purification.

Step c:

[0014] Preparation of the ester of isovaleryl (R)(-)-camitine chloride with (R,S)(±) beta-hydroxybutyric acid benzyl ester 3.

[0015] Acid chloride of isovaleryl (R)(-)-carnitine chloride of step b (0.035 moles) was dissolved in 25 ml anhydrous tetrahydrofurane.

[0016] (R,S)(±) beta-hydroxybutyric acid benzyl ester (7 g; 0.035 moles) of step a was added to the solution.

[0017] The resulting reaction mixture was kept at 25°C under stirring overnight, then ethyl ether was added till complete precipitation. The solid product thus obtained was filtered and washed with ethyl ether. 14 g of product 3 were obtained. Yield 89%.

NMR D<sub>2</sub>O Å 5.7(5H,m,benzyl); 5.5(1H,m,-CH);
5.2(1H,m,COCCH); 5.0(2H,s,CH<sub>2</sub>-benz.)
3.8(2H,m,NCH<sub>2</sub>); 3.2(9H,s,(CH<sub>3</sub>)<sub>3</sub>N+);
2.8-2.5(4H,dd,<u>CH<sub>2</sub>-COOCH;CH<sub>2</sub>COO</u>) 2.2(2H,d,OCOCH<sub>2</sub>)

CH<sub>3</sub>
1.8(1H,m, <u>CH</u>
1; 1.2(3H,d,CH-<u>CH<sub>3</sub></u>); 0.8(6H,d,CH
CH<sub>3</sub>

Step d:

[0018] Preparation of the ester of isovaleryl (R)(-)-carnitine chloride with (R,S)(±)-beta hydroxybutyric acid.

[0019] The product of step c (14 g; 0.031 moles) was dissolved in  $H_2O$  - ethanol (1:1) (1000 ml) and then hydrogenated in the presence of 1.5 g 10% Pd/C at the pressure of 4 atmospheres for 2 hours.

[0020] The reaction mixture was filtered, concentrated to dryness under vacuum and the residue was crystallized from acetone-ethyl ether giving 10 g of a hygroscopic product.

TLC chloroform 4.2 Isopr OH 0.7 MeOH 2.8 H<sub>2</sub>O 1

AcOH 1.1 R<sub>F</sub>: 0.7

[α] #8

= -21(c = 1, H<sub>2</sub>O).

HPLC
Column μBondapack - C1
Eluant KH<sub>2</sub>PO<sub>4</sub>0.05 M - CH<sub>5</sub>CN (85 -15)
UV detector λ = 205

Flowrate 1ml/min

Rt = 14-16 (the diastereomers are shown)

E.A. = $C_{15}H_{30}NO_6CI$		С	Н	N
	calc.	50.6	8.4	3.9
	found	48.93	8.36	3.49

## Example 2

[0021] Preparation of the ester of isobutyryl (R)(-)-carnitine chloride with (R,S)(±)beta-hydroxybutyric acid (ST 730)

Step a:

[0022] same as in Example 1

Step b:

[0023] same as in Example 1, except that isovaleryl (R)(-)-carnitine chloride was substituted by isobutyryl (R)(-)-carnitine chloride

Step c:

[0024] The intermediate 3, ester of isobutyryl (R)(-)-carnitine chloride with (R,S)( $\pm$ )-beta-hydroxybutyric acid benzyl ester was purified via  $\delta$  prep 300 preparative HPLC.

Column prepak C1

Eluant H<sub>2</sub>O-CH<sub>3</sub>CN 70-30 Flowrate 20 ml/min

Yield 50%

NMR D<sub>2</sub>O 
$$\delta$$
 7.5(5H,s,aromatic); 5.8(1H,m,-CH--); 5.3(m,1H,COOCH);

5.1(2H,s,CH<sub>2</sub>benz.); 4.0-3.8(2H,m,NCH<sub>2</sub>); 3.2(9H,s(CH<sub>3</sub>)<sub>3</sub>N); 2.8(2H,m,

 $CH_2COO)$ ; 2.6(2H,m, $CH_2COOH)$ ; 1.8(1H,m,OCOCH); 1.3(3H,d, $CH-CH_3$ );



analytic HPLC
Column μ Bondapak C<sub>1</sub>
Eluant phosphate buffer 0.05M-CH<sub>2</sub>CN 60-40
Flowrate 1 ml/min UV detector λ = 205 nm
Rt = 10.75



```
[0025] Ester of isobutyryl-(R)(-)-carnitine with (R,S)(±)beta-hydroxybutyric acid (ST 730).
[0026] Same as step d of Example 1.
[a]
= -20.3(C = 1H<sub>2</sub>O).
   TLC CHCl3 - H2O-IsopOH - MeOH -AcOH
         (4.2 - 1.05 - 0.7 - 2.8 - 1.05)
   NMR DoO $5.7 (1H.m.-CH-); 5.25(1H.m.COOCH); 3.9-3.7(2H.m.N+CHo);
                                                        CH<sub>3</sub>
                              oco
    3.2(9H,s(CH_2)_2N^+); 2.9(1H,m,CH_2COO); 2.7(2H,m,CH_2COOH);
    C14H28NO6CI
                                                  Н
                                                                         CI
                      caic.
                                  49.19
                                                  8.25
                                                            4.10
                                                                          10.04
                      found
                                  50.26
                                                  8.12
                                                            3.59
                                                                          10.61
HPLC
Column µ Bondapack - C1
Eluant KH2PO40.05M - CH3CN 85-15
UV detector \lambda = 205 \text{ nm}
Flowrate 1 ml/min
Rt = 8.10 - 9.98 (the two diastereomers are thus shown)
```

## Example 3

[0027] Preparation of the ester of acetyl (R)(-)-carnitine chloride with (R,S)(±)beta-hydroxybutyric acid (ST 765)

Step a:

[0028] same as in Example 1

Step b:

[0029] same as in Example 1, except that isovaleryl (R)(-)-carnitine chloride was substituted by acetyl (R)(-)-carnitine chloride.

Step c:

[0030] Intermediate 3, ester of acetyl (R)( -)camitine chloride with (R,S)(±)-beta-hydroxybutyric acid benzyl ester, was purified via preparative HPLC as described in step C of Example 2. Yield 50%

NMR D<sub>2</sub>O Ô7.5(5H,s,aromatic); 5.7(1H,m,-CH-); 5.4-5.0(3H,m,s,COOCH-,

0C0

CH2-Ar); 3.8(2H,m,N+CH2); 3.2(9H,s,(CH3)3N+); 2.8-2.5(4H,m,COOCH2,

CH2COOH) 2.2(3H,s.COCH3); 1.4(3H,d,CHCH3)

analytic HPLC

Column µ Bondapack C1

Eluarit phosphate buffer KH2PO4 0.05M - CH3CN 60-40

Flowrate 1 ml/min UV detector λ = 205 nm

Rt = 11.73

Step d:

[0031] Ester of acetyl (R)(-)-carnitine chloride with (R,S)(±)beta-hydroxybutyric acid (ST 765)

[0032] Prepared as described in step d of Example 1

[a]

ā

= -22.9 (H<sub>2</sub>O 1.2%)

TLC CHCl3 - H2O - isoprOH - MetOH - AcOH (4.2 - 1.05 - 0.7 - 2.8 - 1.05)

Rf = 0.6

NMR D<sub>2</sub>O  $\delta$  5.7(1H,m,-CH-); 5.3(1H,m,COOCH); 3.9-3.7(2H,m,N+-CH<sub>2</sub>);

OCO

3.2(9H,s,(CH<sub>3</sub>)<sub>3</sub>N+); 2.9(2H,m,CH<sub>2</sub>COO); 2.7(2H,m,CH<sub>2</sub>COOH)

2.2(3H,s,COCH<sub>3</sub>); 1.4(3H,d,CHCH<sub>3</sub>)

E.A. C <sub>13</sub> H <sub>24</sub> NO <sub>6</sub> Cl		С	н	N	CI
	calc.	47.90	7.42	4.29	10.88
	found	47.14	7.57	4.88	10.64

H<sub>2</sub>O 0 46%

HPLC

Column µ Bondapack C1

Eluant phosphate buffer KH2PO4 0.05M - CH3CN 90 - 10

Flowrate 0.5 ml/min

UV detector  $\lambda = 205 \text{ nm}$ 

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Rt = 11.68 - 12.83 (the two diastereomers are thus shown).
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Example 4

[0033] Preparation of the ester of propionyl (R)(-)-carnitine chloride with (R,S)(±)beta-hydroxybutyric acid (ST 780).

Step a:

[0034] same as in Example 1

Step b:

[0035] same as in Example 1, except that isovaleryl R(-)carnitine chloride was substituted by propionyl R(-)carnitine chloride.

Step c:

[0036] Intermediate 3, ester of propionyl (R)(-)-carnitine chloride with (R,S)(±)-beta-hydroxybutyric acid benzyl ester, was purified via preparative HPLC as described in step c of example 2.

5.1(2H,s,CH<sub>2</sub>-Ar);4.0(2H,m,N+CH<sub>2</sub>); 3.4(3H,s,(CH<sub>2</sub>)<sub>3</sub>N+); 2.9-2.5(4H,mCH<sub>2</sub>CO

OCH: CH2COOH); 2.3(2H.t.OCOCH2); 1.4-1.0(6H.m.CH2CH3; CH3CH)

analytic HPLC
Column μ Bondapack C<sub>1</sub>
Eluant phosphate buffer 0.005 M 60
Acetonitrile 40
Flowrate 1 ml/min
UV detector λ = 205 nm
Rt = 8.46

Step d:

[0037] Ester of propionyl-(R)(-)-carnitine chloride with (R,S)(±)beta-hydroxybutyric acid (ST 780).

[0038] Prepared as described in step d of Example 1.

[α]

= -23.9 (C = 1% H<sub>2</sub>O)

HPLC

Column µ Bondapack C<sub>1</sub>

Eluant phosphate buffer KH2PO4 0.005 M 90

CH₃CN 10

Flowrate 0.5 ml/min UV detector  $\lambda = 20.5$  nm

Rt = 6.40-7.07 (the two diastereomers are thus shown).

### Example 5

[0039] Preparation of the ester of R(-)-carnitine chloride with R,S(±)beta-hydroxybutyric acid (ST 784).

[0040] The compound was prepared as described in the previous Examples 1-4.

25

= -11.1 (C = 1% H<sub>2</sub>O)

HPLC Column Novapak C<sub>1</sub> mobile phase KH<sub>2</sub>PO<sub>4</sub>50 mM

Flowrate 1 ml/min Rt = 4.56-5.01 min (the two diastereomers are thus shown)

#### Example 6

[0041] Ester of isobutyryl (R)(-)-carnitine chloride with R(-)-beta-hydroxybutyric acid (ST 863).

[0042] The compound was prepared as described in Example 2 (ST 730)

[0043] The compound of step c, ester of isobutyryl (R)(-)-carnitine with R(-)-beta-hydroxybutyric acid benzyl ester, showed the following characteristics:
[6]

(X.)

- - 11.1 (C = 1% MetOH)

HPLC

Column µ Bondapack C1

mobile phase NaClO<sub>4</sub>0.05M-CH<sub>3</sub>CN (60-40)

```
Flowrate 1.5 ml/min
UV detector λ = 205 nm
R_{\star} = 15,64 \text{ min}
[0044] The compound of step d, i.e. the title compound ester of isobutyryl (R)(-)-carnitine chloride with R(-)-beta-hydroxybutyric acid (ST
863), showed the following characteristics:
25
= -11.6 (C = 1% HzO)
HPLC
Column u Bondapack C1
mobile phase KH2PO40.05M - CH3CN 70-30
Flowrate 1 ml/min
UV detector λ = 205 nm
R_* = 8.25
Example 7
[0045] Ester of isobutyryl (R)( -)-carnitine chloride with S(+)-beta-hydroxybutyric acid (ST 864).
[0046] The compound was prepared as described in Example 2 (ST 730)
[0047] The compound of step c, ester of isobutyryl (R)(-)-carnitine chloride with S(+)-beta-hydroxybutyric acid benzyl ester, showed the
following characteristics:
[a]
= -15.4 (C = 1% MetOH)
HPLC
Column µ Bondapak C1
mobile phase NaClO<sub>4</sub>0.05M-CH<sub>3</sub>CN (60-40)
Flowrate 1.5 ml/min
UV detector λ = 205 nm
Rt = 14.79 \min
[0048] The compound of step d. i.e. the title compound ester of isobutyryl (R)(-)-carnitine with S(+)-beta-hydroxybutyric acid (ST 864).
showed the following characteristics:
25
= - 21.7 (C = 1% H<sub>2</sub>O)
HPLC
Column µ Bondapack C1
mobile phase KH<sub>2</sub>PO<sub>4</sub>0.05M-CH<sub>3</sub>CN (70-30)
Flowrate 1 ml/min
UV detector λ = 205 nm
R_* = 7.32 \text{ min}
Example 8
[0049] Ester of butyryl (R)(-)-carnitine chloride with (R,S)(±)-betahydroxybutyric acid (ST 877).
[0050] The compound was prepared as described in Example 1.
[0051] The compound of step c, ester of butyryl (R)(-)-carnitine chloride with (R,S)(±)-betahydroxybutyric acid benzyl ester, showed the
following characteristics:
```

25 0

```
= -12.8 (C = 1% HzO)

HPLC

Column 53 ODS1 (100 mm x 1 mm) Sperisorb

mobile phase KHzPOz0.05M - CHzCN 70-30

UV detector \( \times = 205 \) nm

Flowrate 0.1 ml/min

Rt = 30 min
```

[0052] The compound of step d, i.e. the title compound ester of butyryl (R)(-)-carnitine chloride with (R,S)(±)-beta-hydroxybutyric acid (ST 877), showed the following characteristics:

[a]

= -18.9 (C = 1% H<sub>2</sub>O)

HPLC Column Bondanak NH2

mobile phase KH2PO40.05M - CH3CN 35-65

UV detector λ = 205 nm Flowrate 0.1 ml/min

Rt = 5.62

NMR D<sub>2</sub>O 
$$\hat{0}$$
 5.6(1H,m CH-); 5.2(1H,m,CH); 3.8(2H,m,N+CH<sub>2</sub>)  
000 CH<sub>3</sub>  
3.2(9H.s.(CH<sub>3</sub>)<sub>3</sub>N+); 2.8(4H,m,CH<sub>2</sub>COO,OCOCH<sub>2</sub>);  
2.4(2H,t,CH<sub>2</sub>COOH); 1.7(2H,m,CH<sub>2</sub>CH<sub>3</sub>);  
1.2(3H,d,CHCH<sub>3</sub>); 1.0(3H,t,CH<sub>2</sub>CH<sub>3</sub>)

Effect of ST 687 on the neurologic deficit, memory impairment and cerebral oedema in post-oligaemic rats.

[9053] The study was conducted with a view to assessing the therapeutical effect of \$T 687 administered i.p. to rats immediately after the effect of a transient forebrain oligonhemia in the experiment animals were detected. In particular, the propensity of post-oligonier rats to develop conditioning was studied in a one-trial passive avoidance task. Concurrently, the neurologic deficit during a 3-day period following the ischaemic instalt was assessed, and finally the extent of the ocderna was assessed by measuring the water content of everbrail tissue.

[0054] The effects of ST 687 were compared with those of acetyl (R)(-)-carnitine studied under identical experimental conditions.

[0055] In the experiments Sprague-Dawley (Iffa Credo) male rats weighing 230-250 g were used, that had been caged (5 rats/cage) under conditions of controlled temperature (22°C=1°C), 50% relative humidity and 12-hours dark-light cycle (light on from 8 a.m. to 8 p.m.). The rats were fed UAR (Epinay Ope, France) laboratory chow and had free access to they water. The rats were caged for 5 before surgery.

[9056] Under light ether anesthesis, carotid arteries were isolated and loosely surrounded by a thread. Twenty-four hours later, reversible incomplete forebrain ischemia was produced by bilateral common carotid artery occlusion combined with softium nitroprasside-induced arterial hypotension (1.1 mg/raf s.c.). Mean arterial blood pressure (MABP) was lowered and maintained nearby 6.6 kPa for 45 min. Then it grandatally returned to normal within the 60th minute when carotid occlusion was removed.

[0057] The neurological deficit was assessed via the observational method described by Irwin S.: Comprehensive observational seasessment: I. A. systematic, qualitative procedure for assessing the behavioural and physiologic states of the mouse. Psychopharmacologia (Berl.), 1968, 13: 222-257, for quantifying the behavioural and physiologic states of the mouse. The rats were lifted vertically by mid-ali approximately 15 cm above a rod and lowered for clicit the visual placing response, usually characterized by an extension of forclimbs before centact. The rating was as follows: 3 = normal behaviour (the rat grasps the rod): 2 = mild anterolaterized rotation of the forelimbs (the grasping reflex only occurs when the rat is placed close to the rod): 1 = severe rotation of the forelimbs and of the body (the grasping reflex occurs occurs when the rat touches the rod): 0 = no grasping reflex. The neurological deficit according to this criterion was evaluated respectively, 3, 24, 48 and 72 hours following oligobatemia.

[9088] The functional aspects of the cerebral ischemic injury were assessed by a one trial learning procedure (passive avoidance reaction) as originally described by Kurz K. H. Pe Perl J. The effects of prior few experience in acquired drive learning. J. Comp. Physiol. Psychol. 1960, 53: 201-206, and more extensively developed by Buresova et al.: Effect of atropine on learning, extinction, retention and retrieval in rats. Psychopharmacologia, 1964, 5: 255-263.

[9059] Four hours after clipping-off untreated post-oligaemic rats were placed into an apparatus consisting of a large illuminated compartment (40 × 40 cm) connected by an opening to a small dark compartment (10 × 10 cm) with an electrified grid floor. The animals placed into the large compartment were allowed to explore the apparatus for three minutes. The latency to enter and the time spent in the small compartment were measured with a stop watch. It beloation to the experimental conditions was repeated 22 and 29 hours after sischemia. At the end of the third habituation trial, the opening between the two compartments was closed and the rat, placed into the small compartment, received intermittent electrical foot-shocks for one minute. The retention of the passive avoidance towards the small compartment was tested 24 and 48 hours after the last habituation trial. i.e. 33 and 77 hours post-oligopatemia, respectively. The criterion used to determine whether an animal was conditioned was based upon the rat remaining in the large illuminated compartment for 180 sec. without entry into the small dark compartment.

[0060] Immediately after the last retention trial, i.e. 77 hours post-oligohaemia, rats were sacrificed by decapitation, their brains rapidly removed and macroscopically examined in terms of swelling. Brain water content was determined by the method wet weight/dry weight.

Death-rate, cerebral cedema and passive avoidance in post-oligaemic rats

and following treatment with acetyl(R)(-)-camitine and ST 637

	Dose mg. kg i.p.	n, ì	n.2	Beath rate after 72 hours		Incidence of retention of canditioned response	
	twice a day			(%)		+ 53 h	+ 77.h
Oligobaemia	g	19	đ	\$2,¢	£,38	44,4	32,3
Acetyl (R)(-)-carni time	12,5	15	ia .	33,3	80,0	80,0	80,0
C.386	25,0	17	10	41,2	50,0	60,0	70,0
ar all all and an area	\$0.3	20	13	\$0,0	90,3	70,0	50,0
	4.5	19		47,4	80.d	20,2	45.9
ST 687	+ 5.3	16	:3	37,5	<b>50.</b> 0	60,0	60.0
	7.5	16	ε	6Z.5	ۋ,7	53.3	65.7

n : • number of oligeemic rats

Protective effect of ST 784 against acetaminophen (paracetamol)-induced hepatic damage.

[0061] Paracetamol has been widely used as analgesic and antipyretic. Paracetamol overdosage is known to provoke serious hepatic damages.

[0062] Male Wisstar rats weighing 200-250 g (15 rats/group) that had been keept fasting for at least 12 hours, were administered a single dose of paracetamod (1 g/gb pdo/ weight, per os). 100 g paracetamod were dissolved in 1000 ml of 5% (w/v) carboxymethylcellulose suspension in water. (Hence, the animals were actually administered 10 ml paracetamol solution/kg body weight). 101 mg ST 784/kg body weight were administered orally (as aqueous solution) 1,8 and 24 hours, respectively, following paracetamol administration. The animals were scriftced 32 hours following paracetamol administration.

[0063] Transaminases (SGOT and SGPT) were measured in blood serum. ST 784 provoked a decrease in transaminases exceding 60% (p≦5) with respect to the control animals.

[9064] The compounds of the present invention are orally or parenterally administered, in any of the usual pharmaceutical forms which are prepared by conventional procedures well-known to those persons skilled in the pharmaceutical technology. These forms include solid and liquid oral unit dosage forms such as tablets, capsules, solution, syrups and the like as well as injectable forms, such as sterile solutions for amountes and thials.

[0065] For these pharmaceutical forms the usual solvents, diluents and excipients are used. Optionally, sweetening, flavouring and

n 2 = number of survivors

n 3 · number of rats exhibiting cerebral pedema

p < 0.35 according to the continuity-corrected chi<sup>2</sup> test.

preservative agents can also be bresent. Non limiting examples of such agents are sodium carboxymethylcellulose, polysorbate, mannitol, sorbitol, starch, avicel, talcum and other agents which will be apparent to those skilled in the pharmaceutical technology.

[0066] The dose which is administered will be determined by the attending physician having regard to the age, weight and general conditions of the patient, utilizing sound professional judgement. Although effective results can be noticed at doses as low as 5 to 8 mg/kg of body weight daily, a dose of from about 10 mg/kg of body weight is preferred. Whenever necessary, larger doses can be safely administered in view of the low toxicity of the compounds of this invention.

[0067] As non-limiting examples and depending on the specific pharmaceutical form of administration, the following dosages can be indicated:

for the phials: from 5 to 500 mg for the capsules: from 15 to 50 mg for the tablets: from 15 to 500 mg for the oral solution: from 15 to 50 mg

#### Claims

Esters of (R)(-)-carnitine and acyl (R)(-)-carnitines with beta-hydroxybutyric acid.
 Esters according to claim 1, in the form of pharmacologically acceptable salts of formula (I)

wherein  $X_{-}$  is the anion of a pharmacologically acceptable acid and R is hydrogen or a straight of branched acyl group having from 2 to 5 carbon atoms

3. Esters according to claim 1, in the form if inner salts (I')

- 4. Esters according to claims 2 or 3, wherein R is selected from hydrogen, acetyl, propionyl, n-butyryl, isobutyryl and isovaleryl.
  5. Esters according to claim 2, wherein X is selected from chloride, bromide, orotate, acid aspartate, acid citrate, acid phosphate, acid funnante, lactate, acid malente, acid oxalate, acid sulfate and glucosephosphate.
- 6. An orally or parenterally administrable composition comprising an ester of formula (I) or (I') as active principle.
- 7. An orally or parenterally administrable composition for inhibiting neuronal degeneration, liver proteolysis and for the treatment of coma comprising an ester of formula (I) or (I') as active principle and a pharmacologically acceptable excipient therefor.
- 8. Composition according to claim 7, in unit dosage form, comprising between about 5 and 500 mg of an ester of formula (I) or (I')

## Claims

1. A process for preparing esters of (R)(-)-carnitine and acyl (R)(-)-carnitine wiht β-hydroxybutyric acid of general formula (I)

wherein X is the anion of a pharmacologically acceptable acid and R is H or a straight or branched acyl group having from 2 to 5 carbon atoms, comprising:

 condensing the sodium salt of β-hydroxybutyric acid with benzyl chloride in the presence of crown-ethers, in an organic solvent, in an inert gas atmosphere, at a temperature comprised between 20°C and 30°C, for 1 -2 hours, and isolating the benzylester of βhydroxybutyric acid thus obtained via distillation under vacuum; 2) condensing the acid chloride of (R)-(-camitine or acyl (R)(-)-camitine with the benzylester of β-hydroxyburyic acid in an inert anhydrous organic solvent, at a temperature comprised between 20°C and 30°C, for 12-24 hours, and isolating the compound thus obtained, acyl (R)(-)-camitine ester with β-hydroxyburyic acid benzylester, from the reaction mixture by precipitation with an organic solvent, such as exhyl ether or accolatificit; and

3) hydrogenating the compound obtained in step 2) in a water or ethanol solution or mixtures thereof, in the presence of a hydrogenation catalyst, such as 5% or 10% Pd/C, for 9-18 bit mixtures, at a pressure of 2-5 hydrogen atmospheres, and isolating the product thus obtained, acyl (R)(-)-carnitine ester with  $\beta$ -hydroxybutyric acid, by concentrating under vacuum the solution to droness.



# EUROPEAN SEARCH REPORT

Application Number

		DERED TO BE RELEVA	NT	
Cutegory	Citation of document with i	ndication, where appropriate, usages	Referent to claim	CLASSIFICATION OF THE APPLICATION (Inc. CL5.)
	DE-A-3015636 (SIGNA TAI	5 S.P.A.)	1-8	C07C219/22
- 1	" the whole document "			A61K31/22
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				TECHNICAL FIELDS SEARCHED (m. C.S.)
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